



# TWO-STAGE ALKALINE AND ACID PRETREATMENT APPLIED TO SUGARCANE BAGASSE TO ENRICH THE CELLULOSIC FRACTION AND IMPROVE ENZYMATIC DIGESTIBILITY

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#### ABSTRACT

Sugarcane bagasse (SB) is made up of cellulose (32-43%), hemicellulose (19-34%) and lignin (14-30%). Due to high recalcitrant nature of SB, pretreatment is required to deconstruct its structure and enrich the cellulosic fraction. A two-stage NaOH and maleic acid pretreatment was applied to SB to enrich its cellulosic fraction. SB used in the present study is composed of cellulose (40.4 wt%), hemicellulose (20.9 wt%), lignin (22.5 wt%) and ash (4.0 wt%). After one-stage NaOH pretreatment, its cellulosic fraction increased to 61.8 wt% and later increased to 80.1 wt% after the second-stage acid pretreatment. Lignin fraction decreased to 3.0 wt% after onestage NaOH pretreatment and remained unaffected after the acid pretreatment step. Hemicellulose fraction decreased substantially after the second-stage pretreatment with maleic acid. Pretreated SB displayed high crystallinity index and improved enzymatic digestibility. Hydrolysates of pretreated SB contained very low amount of xylose and subsequent fermentation by Saccharomyces cerevisiae -IQAr/45-1 resulted to ethanol level of 8.94 g/L. Maximal ethanol yield of 0.49 g/g (95.8% of theoretical yield) and productivity of 0.28 g/L/h was attained. At the same time, biomass yield and productivity of 0.47 g/g and 0.27 g/L/h respectively were obtained. Two-stage NaOH and maleic acid pretreatment led to ~ two-fold increase in cellulosic fraction and enhanced the enzymatic digestibility of SB up to 70.4%. The resulted enzymatic hydrolysate was efficiently utilized by S. cerevisiae -IQAr/45-1 to produce high yield of ethanol. Thus, optimization of enzymatic hydrolysis at low enzyme loading is expected to further improve the process and reduce cost.

# **1. INTRODUCTION**

Sugarcane bagasse (SB) is an important feedstock for second generation bioethanol production due to its large abundance, non-competitiveness with food or feed requirement, easy transportation and rich in accessible carbohydrates (Singh et al. 2015; Chandel et al., 2012). The composition and productivity of bagasse is dependent on sugarcane variety, climate, location, plant age, and soil types (Zhao and Li, 2015). SB is made up of cellulose (32-43%), hemicellulose (19-34%) and lignin (14-30%) (Brienzo et al., 2016; Timung et al., 2016).

Cellulose exists as D-glucose subunits, linked by  $\beta$ -1,4 glycosidic bonds (Jönsson et al., 2016) and its microfibrils are chemically bound to lignin and hemicellulose (Zhang et al., 2016). The cellulose in a plant consists of parts with a crystalline (organized) structure, and parts with not well-organized, amorphous structure (Rongpipi et al., 2019).

Hemicelluloses, unlike cellulose, are polymers constituted of heterogeneous, branched polysaccharide composed of C<sub>5</sub> sugars (xylose, arabinose) and C<sub>6</sub> sugars (mannose, glucose and galactose), which serve as a connection between lignin and the cellulose fibers (Lee et al., 2014). Xylan is a dominant component of hemicellulose from hardwood and agricultural plants, such as grasses and straw while for softwood it is glucomannan (Álvarez et al., 2016). Lignin is a complex, cross-linked, three-dimensional polymers of phenolic monomers having both aliphatic and aromatic constituents (Karunarathna and Smith, 2020). Lignin is a constituent that is known to inhibit enzymatic saccharification and fermentative microorganisms (Kucharska et al., 2020). Nevertheless, the matrix structure of lignocellulosic biomass prevents the enzymatic saccharifcation and subsequently sugars fermentation to bioeth-



Detritus / Volume 13 - 2020 / pages 106-113 https://doi.org/10.31025/2611-4135/2020.14005 © 2019 Cisa Publisher. Open access article under CC BY-NC-ND license anol (Sun et al., 2015). Thus, pretreatment is required to deconstruct the intact structure by removal of hemicellulose/lignin and improve the enzyme accessibility to cellulose (Isaac et al., 2018; Acharjee et al., 2017). On the other hand, pretreatment can only be considered effective if it minimizes carbohydrate degradation and the production of enzyme inhibitors as well as toxic products for fermenting microorganisms (Thite et al., 2019; Laluce et al., 2019a).

Research activities in the last few years have been directed towards improving sugar yield from lignocellulosic biomass through the application of physical, chemical, physicochemical and biological pretreatments (Mohapatra et al., 2017). Chemical pretreatments with dilute inorganic acids and alkali have been applied to hydrolyze lignocellulosic biomass (Laluce et al., 2019b; Matsakas et al., 2018; Silveira et al., 2015). On the other hand, dicarboxylic organic acids such as maleic acid are considered alternative to inorganic acids (e.g. sulfuric acid) because they have lower hazardous properties and lower inhibitory by-products generation as well as the ability to hydrolyze  $\beta$ -(1,4) glycosidic bonds (Girolamo et al., 2016; Jung et al., 2014). Maleic acid pretreatment of oil palm trunk was reported to effectively remove hemicellulose and part of acid soluble lignin (Qiao et. at. 2019).

The combinations of delignification process with dilute acid pretreatment have been applied to remove lignin and solubilize most of the hemicellulose sugars from SB (Zhang et al., 2018; Chandel et al., 2014). However, pretreatment combining NaOH pretreatment (first-stage) and maleic acid pretreatment (second-stage) is scarcely reported in the literature.

Thus, a two-stage pretreatment of SB was proposed in this study to degrade lignin using 3.0% NaOH at 121°C for 60 min in the first-stage and then to solubilize hemicellulose using 0.3% maleic pretreatment at 121°C for 60 min in the second-stage. Subsequently, the one-stage and two-stage pretreated SB was characterized by chemical constituent analysis and X-ray diffraction (XRD) was used to investigate the structural changes that occurred during pretreatments. Furthermore, enzymatic saccharification of pretreated SB and ethanol fermentation performance by *Saccharomyces cerevisiae* (IQAr/45-1) were evaluated.

# 2. EXPERIMENTAL

# 2.1 Sample preparation and processing

Sugarcane bagasse was collected from a local sugar mill plant, Santa Cruz (member of São Martinho group) located in Américo Brasiliense, São Paulo, Brazil and transported at low temperature. Bagasse was stored in a laboratory freezer at -20°C. Frozen samples were defrosted and dried at 60°C in a laboratory incubator until a constant weight was obtained (<10 % wt, moisture). Dry bagasse samples were stored in transparent plastic bags at room temperature before use.

Milling the dry sugarcane bagasse (SB) by a physical method initiated the degradation of lignocellulose and conversion of the biomass fibers into particles, thereby increasing biomass surface area (Palmowski and Müller, 2000). For this, sugarcane bagasse was ground for 15 min/cycle in Marconi Ball Mill with Closed Chamber (model MA350) to obtain particles of  $\leq 0.5$  mm after passing through a set of superposed sieves of different meshes (32, 35, and 150 mesh).

Subsequently, SB was refluxed in a Soxhlet extraction apparatus containing a mixture of toluene:ethanol (2:1, v/v) (Sun et al., 2004). The sample was refluxed in water for 30 min to remove the remaining solvents before being dried in a Biochemical Oxygen Demand (BOD) incubator, as recommended in the literature (Binod et al., 2012) at 60°C to constant weight before being stored in a desiccator at room temperature until use.

#### 2.2 Pretreatment with alkaline solution (one-stage)

Dry sugarcane bagasse (particle size  $\leq 0.5$  mm) was treated with different concentrations of NaOH solution (0.5%-3%, w/v) in 125 ml Erlenmeyer flasks at solid/liquid ratio of 1:20. Flasks were gently swirled to enable the solid to become completely soaked in the solution before been transferred into the autoclave and heated for 60 minutes at 121°C. After heating, samples were removed from the autoclave and allowed to cool before filtration under vacuum using Whatman No. 1 filter papers.

Filtrate was collected and used to assay for glucose and xylose while the insoluble material (residues) on the filter paper was washed several times with deionized water to neutral pH and dried at 60°C to a constant weight. Dry residues were used for compositional analysis and dilute acid pretreatment step.

### 2.3 Selection of optimum condition for acid pretreatment step

Dilute acid solutions of maleic acid (MA) in the range of 0.1%-0.5% (w/v) were used to hydrolyze sugarcane bagasse at 121°C (autoclave) for 60 minutes with solid/liquid ratio of 1:20. Sample was removed after autoclaving and allowed to cool on ice bath and filtered under vacuum using Whatman No. 1 filter papers. Filtrate was collected and used to measure the soluble reducing sugars (SRS) using 3,5-Dinitrosalicylic acid (DNS) reagent. The residue on the filter paper was washed with deionized water to neutral pH and dried in a hot air oven at 60°C till the weight remained constant and measured. The lowest concentration of acid that yielded the highest amount of SRS and highest biomass loss\_was selected for the second step pretreatment.

The amount of released SRS in pretreatment filtrate from SCB was measured using D-glucose as standard and expressed as % released SRS as mentioned below.

$$\% released SRS = \tag{1}$$

= 
$$\frac{reducing \ sugar \ released \ in \ filtrate \ during \ pretreatment \ (mg)}{biomass \ used \ for \ pretreatment \ (mg)}$$

0,

Decrease in weight of dry biomass was calculated as mentioned below and expressed as % weight loss in biomass.

% weight loss in biomass =  $\frac{biomass \ obtained \ after \ drying \ (mg)}{biomass \ used \ for \ pretreatment \ (mg)} \times 100$  (2)

# 2.4 Two-stage alkaline and acid pretreatments

Dry residues which originated from the one-stage pretreatment with different concentrations of NaOH were sus-

 $\times 100$ 

pended in dilute solution of maleic acid (0.3%, w/v) in 125 ml Erlenmeyer flask at solid/liquid ratio of 1:20. Flasks were placed in autoclave and heated for 60 minutes at 121°C. Thereafter, flasks were removed and allowed to cool before filtration. Filtrate was collected and used to assay for glucose and xylose, while the residues on the filter paper was washed several times with deionized water to neutral pH before been dried in BOD at 60°C to a constant weight and used for chemical composition analysis.

# 2.5 Determination of the chemical composition of untreated SB and pretreated fractions

The chemical components of untreated and pretreated SB were analyzed using the method described by National Renewable Energy Laboratory (Sluiter et al., 2008). 300 mg dry residues resulting from pretreatment or untreated SB was weighed into a pressure glass tube and 3.0 ml of 72% sulfuric acid was added, the mixture was stirred with glass rod and tube was place in water bath at 30°C for 1 h. The mixture was stirred while in water bath at every 10 min. After 1 h of hydrolysis, tube was removed from water bath and 84 ml of deionized water was added to bring the acid concentration to 4%. The tube was covered and inverted several times to allow sample to mix and autoclaved at 121°C for 1 h.

After autoclaving, the mixture was cooled down to room temperature and separated into solid and liquid fractions by vacuum filtration using Whatman No.1 filter papers. Solid fraction was then dried at 105°C in order to measure acid insoluble residue (AIR). Acid insoluble residue was burnt in muffle furnace at 600°C for about 6 h to measure acid insoluble lignin (AIL) content as the difference between AIR and ash. Before neutralization, the absorbance of liquid fraction at 240 nm was measured by UV-visible spectrophotometer (Cirrus 80 ST, Femto, São Paulo, Brazil) to determine acid soluble lignin (ASL). Liquid fraction was neutralized with calcium carbonate and the resulting filtrate was used to assay for glucose and xylose.

Glucose was assayed using a commercial enzymatic kit (GOD-PAP, Laborlab) and xylose by the phloroglucinol method using xylose as standard (Ebert et al., 1979). Total furans were estimated by a spectrophotometric method based on the difference in absorbance at 284 and 320 nm (Martinez et al., 2000) using a UV/Vis/NIR-spectrometer with a 3D WB Detector Module (Perkin Elmer, Inc., Shelton, CT USA). A standard curve was obtained for each assay by linear regression using the software OriginPro 8 from OriginLab Corporation.

#### 2.6 X-Ray Diffraction (XRD)

XRD was used as a tool to investigate the structural changes that occurred during pretreatments of sugarcane bagasse. The crystalline structure of biomass is mainly due to the strong hydrogen bonding of cellulose chains and Vander Waals forces of glucose molecules in the cellulose (Naik et. al., 2010; Sasmal et. al., 2012).

The crystallinity of the cellulose fiber was evaluated by X-Ray Diffraction (Siemens D5000 diffractometer, Munich, Germany). Copper Karadiation, 30.0 kV of voltage and 15

mA of electric current, and a rate of 2.0 degrees per minute for a 20 continuous scan from 5.0 to 50.0 degrees were applied. This analysis allowed the detection of the amorphous part of the lignocellulosic biomass, as well as the modification of the crystalline structure of the cellulose.

The crystallinity index (CI) was obtained from the ratio of the maximum peak intensity 002 ( $I_{002}$ , 20 = 22.5) and minimal depression ( $I_{am}$  20 = 18.5) between peaks 001 and 002 (Segal et al., 1959; Rodrigues et al., 2007) as mentioned below.

$$CI(\%) = \frac{I_{002} - I_{am}}{I_{002}} \times 100$$
(3)

where  $I_{_{002}}$  is the maximum intensity of the 002 peak and  $I_{_{am}}$  the minimal depression of the amorphous structure.

#### 2.7 Enzymatic hydrolysis

Dry pretreated SB and untreated SB were each soaked in 50 mM sodium citrate buffer (pH 4.8) at solid loading of 5% (w/v) in 125 ml Erlenmeyer flask. Flask was placed in incubator (Tecnal TE-391, Piracicaba, SP, Brazil) and incubated for 2 h at 50°C with shaking speed of 150 rpm. Thereafter, the soaked sample was supplemented with 8.7 FPU/g dried sample of Cellulase from Trichoderma reesei, (Sigma Aldrich) and 5.6 IU/g dry sample of  $\beta$ -glucosidase (Sigma Aldrich). A dose of 0.005% sodium azide was introduced to avoid any microbial contamination and 1.0% (v/v) Tween 80 was added to facilitate the enzymatic action. Enzymatic hydrolysis was performed at 50°C for 72 h with shaking at 150 rpm. Samples were withdrawn at 6 h, 12 h, 24 h, 48 h and 72 h intervals and enzymes were inactivated by boiling at 100°C for 10 min after which samples were cooled on ice before subsequently analyzed for glucose released using a commercial enzymatic kit (GOD-PAP, Laborlab).

Saccharification (%) was calculated as mentioned below:

% saccharification =

(4)

=  $\frac{\text{reducing sugar released by enzymatic hydrolysis (mg)}}{\text{initial solid biomass used for hydrolysis (mg)}} \times 100$ 

#### 2.8 Fermentation

The Saccharomyces cerevisiae -IQAr/45-1 is a thermotolerant ethanologenic yeast strain obtained from the hybridization between parental strains of *S. cerevisiae* and three Brazilian industrial strains (PE-2, CAT-1, SA-1) during fermentation of non-sterilized molasses (Laluce et. al. 2013). This strain can only ferment hexose sugars and it was maintained in medium containing (g/L): glucose, 30.0; yeast extract, 3.0; peptone, 5.0; agar, 20.0 at pH 6.0  $\pm$  0.2 and temperature 30°C. Starter culture was developed by growing the cells at 30°C for 24 h in a culture medium containing (g/L): glucose, 30.0; yeasts extract, 3.0; peptone, 5.0; pH 6.0  $\pm$  0.2.

The fermentation of enzymatic hydrolysates of twostage NaOH+MA pretreated SB (17.3 g/L glucose, pH 6.0  $\pm$  0.2) was carried out in 125 ml Erlenmeyer flask with a working volume of 50 ml supplemented with 3 g/L of yeast extract and 5 g/L of peptone. It was inoculated with *S. cerevisiae* (10.0% v/v) at optical density (OD<sub>600</sub>) of 0.6. Sample was incubated at 30°C for 30 h with shaking at 150 rpm. Samples were centrifuged at 10,000 g for 15 min at 4°C and the cell free supernatant was used to determine the ethanol and residual sugar concentration.

Cell concentrations were measured at optical density of 600 nm and related to the cell dry weight through a calibration curve. Ethanol was estimated using acidified potassium dichromate solution as described by Caputi et al., (1968).

Fermentation parameters were calculated as mentioned below:

$$Y_{P/S} = \frac{E_f - E_i}{(S_i - S_f)}$$
(5)

 $Y_{P/S}$  is the ethanol yield,  $E_i$  and  $E_f$  are the ethanol concentration at the beginning of the fermentation and the end of the fermentation (g/L) respectively; while  $S_i$  and  $S_f$  are the total sugar concentration at the beginning of the fermentation and the end of the fermentation (g/L), respectively.

$$Q_{P_P} = \frac{E_{f-E_i}}{t_f - t_i} \tag{6}$$

 $Q_{pp}$  is the volumetric ethanol productivity (g/L/h),  $E_i$  and  $E_f$  are the ethanol concentration at the beginning and end of fermentation (g/L), respectively; while  $t_i$  and  $t_f$  are the fermentation time at the beginning of the fermentation and the end of the fermentation (g/L), respectively.

$$Y_{X/S} = \frac{X_f - X_i}{S_i - S_f} \tag{7}$$

 $Y_{x/S}$  is the biomass yield  $X_i$  and  $X_f$  are the biomass concentration at the beginning of the fermentation and the end of the fermentation (g/L), respectively; while  $S_i$  and  $S_f$  are the total sugar concentration at the beginning of the fermentation and the end of the fermentation (g/L), respectively.

$$Q_{P_X} = \frac{x_{f-X_i}}{t_{f-t_i}}$$
(8)

 $Q_{Px}$  is the biomass productivity (g/L/h), X<sub>i</sub> and X<sub>f</sub> are the biomass concentration at the beginning and end of fermentation (g/L), respectively; while t<sub>0</sub> and t<sub>r</sub> are the fermentation time at the beginning of the fermentation and the end of the fermentation (g/L), respectively.

Fermentation efficiency (%) = 
$$\frac{Actual Yield}{Theoretical Yield} \times 100$$
 (9)

where theoretical yield is equivalent to 0.511 g/g.

#### 2.9 Data analysis

The graphs were created using the software OriginPro 8 from OriginLab Corporation. Each data was expressed as a mean of standard deviation (SD) of triplicate measurements.

# 3. RESULTS AND DISCUSSION

### 3.1 Pretreatments

A preliminary study to select the optimal condition for acid pretreated step is presented in Table 1. One-stage maleic acid pretreatment of SB resulted to loss of dry weight ranging from 10.6% to 23.4%, while the yield of SRS in the acid hydrolysate ranges from 8.5% to 10.6%. The loss of dry weight was mainly attributed to hemicellulose solubilization during acid pretreatment (Baruah et al., 2018, Li et al., 2016). Thus, the yield of SRS and loss of dry weight were directly correlated with acid concentration. The mass differences between loss of dry weight and yield of SRS could be connected to the presence of other extractive components in the SB, which were solubilized during acid pretreatment.

Table 2 shows the results of one-stage NaOH and twostage NaOH and maleic acid pretreatments applied to SB. It was found that increases in NaOH concentration from 0.5% to 3.0% led to corresponding increases in cellulosic fractions of SB from 45.5 wt% to 61.8 wt%. The proportionality observed between NaOH concentration and cellulosic fraction was due to lignin removal at increasing NaOH concentration. According to the literature, alkaline pretreatment can facilitate dissociation of entire cell wall polymers by breaking hydrogen and covalent bonds thereby enabling effective lignin removal (Thite et al., 2019; Rezende et al., 2011).

The second-stage pretreatment with 0.3% maleic acid led to increase in cellulosic fractions from 46.7 to 80.1 wt%. This increase could be attributed to hemicellulose removal during the acid pretreatment step. Dilute acid pretreatments have been reported to cause hemicellulose solubilization and cellulose enrichment (Zhang et al., 2018; Rezende et al., 2011).

Interestingly, cellulosic fractions of SB resulting from two-stage NaOH and maleic acid pretreatments are connected to the cellulosic fractions from the first-stage NaOH pretreatment. This implies that the higher the cellulosic fraction after first-stage NaOH pretreatment, the higher the cellulosic fraction after the second-stage acid pretreatment. The correlations between first- and second-stage pretreatments were mainly due to hemicellulose solubilization during the second-stage acid pretreatment.

# 3.2 Chemical composition of untreated (raw) and pretreated sugarcane bagasse.

SB used in the present study consists of ~ 10% moisture and particle size of about 0.50 mm. Its chemical composition is made up of cellulose (40.4 wt%), hemicellulose (20.9 wt%), lignin (22.5 wt%) and ash (4.0 wt%) as shown in Table 3. These values are similar to those reported by other authors (Rabelo et al., 2009; Sporck et al., 2017). Furthermore, chemical composition and productivity of SB are mostly dependent on sugarcane variety, climate, location, plant age and soil types (Zhao and Li, 2015).

**TABLE 1:** Effect of maleic acid concentration on the release of soluble reducing sugars (%) in pretreatment filtrate and loss of dry weight (%) during the pretreatment of sugarcane bagasse in autoclave (121°C) for 60 min.

Maleic acid (%, w/v)	Temperature (°C)	Time (min)	Soluble reduc- ing sugars (%)	Loss of dry weight (%)
0.1	121	60	8.5 ± 0.4	10.6 ± 0.4
0.2			10.0 ± 0.6	13.5 ± 0.6
0.3			15.6 ± 0.3	19.1 ± 0.5
0.4			16.9 ± 1.2	20.0 ± 0.3
0.5			19.2 ± 1.1	21.9 ± 0.4
0.6			20.8 ± 0.7	23.4 ± 0.5

TABLE 2: Cellulosic fractions resulting from first-stage NaOH pretreatment and second-stage pretreatment with 0.3% maleic acid in autoclave.

Pretreatment									
First-stage			Second-stage						
NaOH (%, w/v)	Temp (°C)	Time (min)	Cellulose (wt%)	Acid (w/v)	Temp (°C)	Time (min)	Cellulose (wt%)		
0.0	121	60	42.7 ± 0.6	0.3% MA	121	60	46.7 ± 0.7		
0.5			45.5 ± 0.6				56.5 ± 1.5		
1.0			48.9 ± 1.0				63.0 ± 1.4		
1.5			55.5 ± 1.0				68.7 ± 1.2		
2.0			58.4 ± 0.3				73.8 ± 1.2		
2.5			59.8 ± 0.8				78.7 ± 0.9		
3.0			61.8 ± 1.3				80.1 ± 1.8		

 TABLE 3: Chemical composition of solid fractions of untreated and pretreated sugarcane bagasse and composition of pretreatment hydrolysates.

Pretreatment (60 min, 121°C)		Chemical composition	Constituents of pretreatment filtrate			
	Cellulose (wt%)	Hemicellulose (wt%)	Lignin (wt%)	Ash (wt%)	Xylose <sup>#</sup> (%, g/g)	Total furans (mg/L)
Untreated SB	40.4 ± 0.9	20.9 ± 0.7	22.5 ± 0.4	4.0 ± 0.0	na*	na*
3% NaOH	61.8 ± 1.3	17.6 ± 0.5	3.0 ± 0.1	2.7 ± 0.1	20.7 ± 1.8	na*
3% NaOH +0.3% MA	80.1 ± 1.8	4.0 ± 0.5	3.7 ± 0.1	1.9 ± 0.2	40.0 ± 1.1	20.6±0.7

Chemical composition of SB varied significantly after pretreatments were applied. Based on the cellulosic fractions obtained after pretreatment, one-stage NaOH (3% NaOH) pretreated SB showed appreciable increase in cellulosic fraction (61.8 wt%) compared to untreated SB (40.4 wt%), while its hemicellulose and lignin fractions decreased to 17.6 wt% and 3.0 wt% respectively. Similarly, the second-stage pretreatment with 0.3% maleic acid showed much significant increase in cellulosic fraction (80.1 wt%), while hemicellulose and lignin fractions decreased to 4.0 wt% and 3.7 wt% respectively. Thus, lignin removal from SB was mainly attributed to NaOH pretreatment step. Delignification of SB by alkaline pretreatments has been reported in the literature (Zhang et al., 2018). Concerning the ash content, one-stage NaOH pretreatment led to significant decrease in ash content (2.7 wt%), while no significant effect on the ash content was obtained after second-stage pretreatment with maleic acid.

Alkaline pretreatment of SB using NaOH led to the removal of major part of lignin fraction with the retention of cellulose and hemicellulose fractions. Conversely, maleic acid pretreatment removed major part of hemicellulose and acid soluble fraction of lignin, while the cellulosic components and insoluble lignin remained largely unaffected. However, pretreatment filtrates from first-stage pretreatment with 3%NaOH and second-stage pretreatment with 0.3% maleic acid contain significant amount of xylose. This suggests that part of hemicellulose was solubilized during both the first-stage NaOH and second-stage maleic acid pretreatments. Furthermore, filtrates from the secondstage maleic acid pretreatment were found to contain very low amount of total furans (20.6 mg/L). The low amount of total furans obtained in this study emphasizes the main benefit associated with the use of organic acid such as maleic acid for pretreatment of SB.

# 3.3 XRD analysis of untreated (raw) and pretreated SB biomass

Lignocellulose crystallinity could be transformed via pretreatment by opening crystal hydrogen bonding, degrading amorphous constituents and increasing crystal regions, thereby affecting subsequent enzymatic saccharification (Zhang et al., 2018). Thus, the XRD patterns and Cl of untreated and pretreated SB were investigated and the results are as shown in Figure 1. The Cl of untreated SB was 55.2% and after pretreatment with 3.0% NaOH, Cl increased to 69.0%, while a higher Cl of 79.2% resulted from the combination of NaOH and maleic acid pretreatment steps.

However, all the pretreated SB presented higher CIs than the control (raw material). This phenomenon was mainly due to the removal of amorphous hemicellulose and lignin. The highest CI value indicates greater removal of hemicelluloses, leaving the crystalline cellulose fraction intact in the pretreated solid residues (Timung et al., 2016). Hence, crystallinity of cellulose was found to increase after pretreatments were applied to SB, mainly due to the increase in cellulose content (Rezende et al., 2011). The CI was found to correlate with the result of chemical composition analysis.

# 3.4 Studies on amenability of the pretreated biomass to enzymatic hydrolysis

In the literature, plenty of reports are available showing cellulose conversion rates of variedly pretreated lignocellulosic biomass after subjecting them to enzymatic hydrolysis.



FIGURE 1: Diffractograms and crystallinity index (CI) of untreated SB and solid fractions resulting from pretreatment with 3.0% NaOH (one-stage pretreatment) and 3.0%NaOH+0.3%MA (onestage NaOH pretreatment followed by the second-stage maleic acid pretreatment) respectively.

Few of them are listed and compared with the results in the present study (Table 4). The lowest conversion rate of 12.9% was obtained from untreated SB (raw material), while the highest conversion rate of 70.4% was obtained from SB emanating from two-stage NaOH and maleic acid pretreatments. The conversion rate of SB (60.2%) emanating from first-stage NaOH pretreatment was significantly higher than the untreated SB. This suggests that the firststage pretreatment with 3% NaOH effectively remove lignin from SB, thereby improving its enzymatic digestibility. On the other hand, the second-stage pretreatment with 0.3% maleic acid was able to remove greater part of hemicellulose, which further enhanced enzymatic digestibility of SB.

The presence of lignin and hemicellulose negatively affects enzymatic hydrolysis by binding to cellulose, thereby impeding its access to cellulose (Zheng et al., 2018; Sabanci et al., 2018). Therefore, the lowest conversion rate



FIGURE 2: Xylose and glucose concentration after 72 h of enzymatic hydrolysis of untreated SB and solid fractions resulting from pretreatment of SB with 3.0% NaOH (one-stage pretreatment) and 3.0%NaOH+0.3% maleic acid (one-stage NaOH pretreatment followed by the second-stage maleic acid pretreatment respectively.

obtained from untreated SB suggests inaccessibility of the enzymes to cellulose due to the presence of lignin and hemicellulose. The enzymes inaccessibility to cellulose appeared to be drastically reduced via pretreatments, hence the improved cellulose conversion rate obtained from the pretreated SB. On the other hand, enzymatic hydrolysates contained low amount of xylose (Figure 2), which can be attributed to the effectiveness of the acid pretreatment step in hemicellulose removal. Also, the presence of xylose in the hydrolysates may be connected to the action of cellulases and  $\beta$ -glucosidase enzymes which caused the disruption of lignocellulosic matrix thereby releasing xylose (Guilherme et al., 2015).

#### **3.5 Fermentation**

Table 5 shows the batch fermentation profile of the S. cerevisiae -IQAr/45-1 on enzymatic hydrolysate of two-stage

the yield of fermentable sugars from sugarcane bagasse after enzymatic saccharification.							
Biomass source	Pretreatment conditions	Type of Enzyme cocktail	Saccharification/ Conversion rate	Reference			
Sugarcane	Raw (untreated)	Cellulase + β- glucosidase	12.9%	This study			

TABLE 4: Comparison of one-stage NaOH pretreatment, two-stage NaOH + maleic acid pretreatment and other pretreatment methods on

Biomass source Pretreatment conditions		Type of Enzyme cocktall	Conversion rate	Reference	
Sugarcane bagasse	Raw (untreated)	Cellulase + β- glucosidase (Novozyme 188)	12.9%	This study	
	3%NaOH (121°C, 60 min)		60.2%	This study	
	Two-stage, 3%NaOH (121°C, 60 min) + 0.3% Maleic acid (121°C, 60 min)		70.4%	This study	
	1% NaOH at 600 W for 4 min, microwave	Commercial cellulase (Zytex)	66.5%	(Binod et al., 2012)	
	1%NaOH (115°C, 20 min)	Primafast 200	38.8%	(Thite et al., 2019)	
	Two-stage, 1%H₂SO₄ (120°C, 30 min) + 0.5% NaOH (120°C, 60 min)	Cellic CTec2	71.4%	(Zhang et al., 2018)	
	Two-stage, 1%H₂SO₄ (120°C, 30 min) + 60% ethanol (120°C, 60 min)		45.9%	(Zhang et al., 2018)	
	1% H <sub>2</sub> SO <sub>4</sub> (115°C, 20 min)	Primafast 200	25.6%	(Thite et al., 2019)	

**TABLE 5:** Fermentation profile of enzymatic hydrolysate of two-stage (3.0%NaOH+0.3%maleic acid) pretreated SB by *Saccharomyces cerevisiae* (IQAr/45-1 strain) after 30 h.

Time (h)	Glucose (g/L)	Ethanol (g/L)	Y <sub>E/S</sub> (g/g)	$Q_{_{Pp}}(g/L/h)$	Biomass (g/L)	Y <sub>x/s</sub> (g/g)	Q <sub>Px</sub> (g/L/h)	E <sub>F</sub> (%)
0	17.3 ± 0.20	0.62 ± 0.03	0.04	0.00	0.17 ± 0.00	0.01	0.00	0.0
30	0.27 ± 0.01	8.94 ± 0.12	0.49	0.28	8.17 ± 0.18	0.47	0.27	95.8

NaOH and maleic acid pretreated SB after 30 h. Glucose was depleted within 30 h of fermentation, by this time the ethanol concentration and ethanol yield have reached 8.94 g/L and 0.49 g/g respectively. Ethanol productivity of 0.28 g/L/h was obtained after 30 h with the corresponding fermentation efficiency of 95.8% based on the theoretical ethanol yield. On the other hand, the yeast cell biomass increased to a maximum level of 8.19 g/L and biomass yield of 0.47 g/g was obtained after 30 h with productivity of 0.27 g/L/h.

High ethanol yield is an indication of efficient conversion of glucose to ethanol by *S. cerevisiae* -IQAr/45-1. The maximum ethanol yield in this study was higher than those found in many literature reports. For example, de Albuquerque Wanderley and co-workers (2013) reported a maximum ethanol yield of 0.40 g/g (equivalent to fermentation efficiency of 78.47%) from batch fermentation of enzymatic hydrolysate of delignified SB by *S. cerevisiae*. Also, Wi et al. (2015) reported ethanol yield of 0.435 g/g (equivalent to 85.0 % of the maximum theoretical ethanol yield) from enzymatic hydrolysates of rice straw pretreated with hydrogen peroxide-acetic acid (HPAC) after fermentation by *S. cerevisiae*.

The high biomass yield obtained in this study could be attributed to the presence of yeast extract and peptone in the fermentation media. According to Chang and Webb, (2017), essential elements (nitrogen, phosphorous and vitamins) need to be supplied to the fermentation media as prerequisite to optimize ethanol yield, since hydrolysate from lignocellulosic biomass is generally regarded to be nutrient-deficient (Lau and Dale, 2009).

# 4. CONCLUSIONS

The effect of one-stage NaOH and two-stage NaOH and maleic acid pretreatments on cellulose enrichment and enzymatic digestibility of SB were studied. Two-stage NaOH and maleic acid pretreatment led to ~ two-fold increase in cellulosic fractions from SB. First-stage NaOH pretreatment efficiently removed lignin fraction, while the second-stage pretreatment with maleic acid efficiently removed hemicellulose fraction. The result of XRD analysis correlated with the changes in the chemical composition of SB as a result of pretreatments.

The presence of low amount of inhibitors in the pretreatment filtrate after the second-stage maleic acid pretreatment highlights the advantage of maleic acid over inorganic acid. Furthermore, two-stage NaOH and maleic acid pretreatment enhanced the enzymatic digestibility of SB up to 70.4%. The enzymatic hydrolysate was efficiently utilized by *S. cerevisiae* -IQAr/45-1 to produce high yield of ethanol. Thus, optimization of enzymatic hydrolysis at low enzyme loading is expected to further improve the process and reduce cost.

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